

Improving the extraction of Ara h 6 (a peanut allergen) from a chocolate-based matrix for immunosensing detection: Influence of time, temperature and additives

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ABSTRACT

The extraction of Ara h 6 (a peanut allergen) from a complex chocolate-based food matrix was optimized by testing different temperatures, extraction times, and the influence of additives (NaCl and skimmed milk powder) in a total of 36 different conditions. Analyses were carried out using an electrochemical immunosensor. Three conditions were selected since they allowed the extraction of the highest levels of Ara h 6. These extractions were performed using 2 g of sample and 20 ml of Tris-HNO₃ (pH = 8) containing: a) 0.1 M NaCl and 2 g of skimmed milk powder at 21°C for 60 min; b) 1 M NaCl and 1 g of skimmed milk powder at 21°C for 60 min; and c) 2 g of skimmed milk powder at 60°C for 60 min. Recoveries were similar or higher than 94.7%. This work highlights the importance to adjust extraction procedures regarding the target analyte and food matrix components.

Keywords:

Peanut
Allergen
Ara h 6
Extraction
Chocolate
Food
Biosensor
Voltammetric

1. Introduction

Food allergy is generally mediated by immunoglobulin E and consists in an immunological hypersensitivity against some proteins or glycoproteins of food. Symptoms can involve the gastrointestinal tract, the skin, and/or the respiratory system. Anaphylactic reactions are rarer but of concern, since they are life threatening (Berin & Sicherer, 2011; Leung & Kamat, 2008). In addition to symptoms' treatment, food allergies can be controlled by avoiding allergen exposure. Thus, sensitive and selective methods to detect allergens in foodstuffs are extremely important: they can clarify cross-contamination situations during food processing or detect "hidden" allergens in food ingredients (Poms, Klein, & Anklam, 2004).

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Peanut is one of the most allergenic foods. Based on the Directive 2000/13/EC (amended by Directives 2003/89/EC and 2007/68/EC), in the European Union, and the Food Allergen Labelling and Consumer Protection Act of 2004 (FALCPA 2004, Public Law 108-282, Title II), in the United States, the presence of peanut in a foodstuff has to be declared. Different peanut allergens have been documented by the World Health Organization/ International Union of Immunological Societies Allergen Nomenclature Sub-Committee (UIS, 2016). Ara h 1 (a vicillin-type 7S globulin) was firstly identified as the major peanut allergen, however, several studies showed that Ara h 2 (2S albumin) presented a higher capacity to release histamine from human basophils containing peanut-specific IgE and a higher skin reactivity in peanut-allergic individuals than Ara h 1 (Koppelman, Wensing, Ertmann, Knulst, & Knol, 2004; Palmer et al., 2005). Recently, Ara h 6 was identified as a relevant allergen since it has also been associated to clinical peanut allergy. Indeed, compared to Ara h 2, it has a similar seroprevalence and equal capacity to bind and release histamine from basophils (Koid et al., 2014; Koppelman et al., 2005). Moreover, Ara h 2 and Ara h 6 are far more stable to peptic digestion than Ara h 1, which can explain why these proteins are more often recognized

by IgE in the sera of peanut-allergic individuals (Koppelman, Hefle, Taylor, & de Jong, 2010). Because of its heat resistance and stability against digestion (Hed, 2009; Koppelman et al., 2010), Ara h 6 can be considered an appropriate marker to detect peanut in food products and industrial production lines.

Different methods for food allergen detection are described, namely, protein-based immunoassays (Monaci & Visconti, 2010; Taylor, Nordlee, Niemann, & Lambrecht, 2009), DNA-based methods (Monaci & Visconti, 2010; Słowianek & Majak, 2011), and mass spectrometry (Monaci & Visconti, 2009). Recently, immuno- and geno-biosensors appeared as interesting alternatives to those methods. They present serious advantages such as reduction of reagents consumption, miniaturization and portability, and reduction of analysis costs (Alves, Barroso, González-García, Oliveira, & Delerue-Matos, 2016), but their application in this field is still in the beginning.

Nonetheless, besides the detection method, a crucial step in the experimental procedure needs to be considered, especially when analysing foodstuffs, due to their intrinsic chemical complexity: the extraction of the target analyte. Often, an additional purification of the target analyte after extraction is needed before analysis. It is, therefore, crucial to assure that during the entire sample preparation the selected analyte is not destroyed or modified. In the particular case of food allergen detection using immunoassays, sample preparation is an essential step, since several conditions can affect the integrity of the target protein, such as the extraction temperature or the pH of the extraction solution. Moreover, matrix components could affect the method's result (e.g. non-specific binding), resulting in false positives. On the other hand, if the target molecule is not correctly extracted, a false negative could be obtained (L'Hocine & Pitre, 2016; Poms, Capelletti, & Anklam, 2004). These are crucial issues, since in some studies, immunosensors have been developed using a purified allergenic protein standard, but they were not applied to real samples, which makes the method's validation incomplete (Huang, Bell, & Suni, 2008; Singh, Sharma, Baltus, & Suni, 2010). Besides this, food samples are usually very complex and different from each other and the allergenic proteins differ in their structure, therefore different extraction conditions can lead to huge variations in the quantified amount. Together with the type of antibodies and the assay used, the kind of extraction can also be one of the main reasons why different results are obtained for the same sample using different immunoassay kits (Jayasena et al., 2015).

Because of the increasing clinical relevance that has been given to Ara h 6, in a previous work we developed and validated the first voltammetric (based on a gold nanoparticle-modified screen-printed carbon electrode) immunosensor for Ara h 6 detection in real food matrices (Alves, Pimentel, Nouws, Correr, et al., 2015). Due to the inherent difficulty of analysing proteins in chocolate (Hurst, Krout, & Burks, 2002; Newsome & Abbott, 1999; Pomés, Vinton, & Chapman, 2004), in the present study, we aimed to investigate the best extraction conditions of Ara h 6 from a complex chocolate-based food matrix, by testing different temperatures and extraction times, as well as the influence of certain additives (NaCl and skimmed milk powder). To the best of our knowledge, this is the first report about the study of the extraction conditions of Ara h 6 for immunoassay detection.

2. Materials and methods

2.1. Chemicals and reagents

Nitric acid ($\geq 65\%$), hydrochloric acid (37%), sodium chloride, β -casein ($\geq 98\%$), 3-indoxyl phosphate (3-IP), magnesium nitrate hexahydrate, tris(hydroxymethyl)aminomethane (Tris)

and streptavidin-alkaline phosphatase (S-AP) from *Streptomyces avidinii* were all obtained from Sigma-Aldrich. The gold ($\text{H[AuCl}_4\text{]}$) standard solution was purchased from Merck and silver nitrate was supplied by Alfa Aesar. The mouse monoclonal anti-Ara h 6 IgG1 antibody (3B8), the biotinylated monoclonal anti-Ara h 6 IgG1 antibody (3E12) and the standard Ara h 6 were acquired from Indoor Biotechnologies. Skimmed milk powder was acquired in a local supermarket.

Ultra-pure water (resistivity = $18.2 \text{ M}\Omega\cdot\text{cm}$) was obtained using a Millipore (Simplicity 185) water purification system.

The solutions of the reagents that were used in the immunoassay were prepared according to Alves, Pimentel, Nouws, Correr, et al. (2015).

2.2. Sensor structure and instrumentation

Screen-printed carbon electrodes (SPCEs, DropSens, Spain), incorporating a conventional three-electrode electrochemical cell ($50 \mu\text{l}$), were used as transducers of the biological reaction. The electrodes' surfaces were nanostructured with gold nanoparticles by placing $0.1 \text{ mM } [\text{AuCl}_4]^-$ (in 0.1 M HCl) on the electrode and applying a current intensity of $-100 \mu\text{A}$ (240 s), followed by a 0.1 V potential (120 s) (Alves, Pimentel, Nouws, Correr, et al., 2015). Afterwards, the modified SPCEs were washed with water, dried with a nitrogen stream and stored in the presence of a desiccant. Before their use, the electrodes were rinsed with the 0.1 M Tris-HNO_3 buffer (pH 7.2).

The electrochemical measurements were performed using a Metrohm Autolab PGSTAT12 potentiostat-galvanostat controlled by GPES4.9 software.

2.3. Immunosensor assay

The immunoassay was performed exactly according to Alves, Pimentel, Nouws, Correr, et al. (2015). Briefly, SPCEs-nAu were coated with a monoclonal anti-Ara h 6 IgG solution ($25 \mu\text{g/ml}$) and left to incubate overnight at 4°C . Surface blocking was performed using a $2\% \text{ (w/v)}$ β -casein solution. The sensor was then incubated with a diluted ($1:250$) sample extract ($40 \mu\text{l}$), an Ara h 6 standard (for calibration) or a Tris-HNO_3 pH 7.2 buffer (for the blank assay). Then, the biotinylated monoclonal anti-Ara h 6 IgG ($1:25,000$) was left to incubate, followed by S-AP ($1 \times 10^{-10} \text{ M}$) addition. The enzymatic reaction was carried out using a mixture of 3-IP and silver nitrate ($1.0 \times 10^{-3} \text{ M}$ and $4.0 \times 10^{-4} \text{ M}$, respectively). After 20 min of reaction, a linear sweep voltammogram was recorded from -0.02 V to $+0.4 \text{ V}$ (50 mV/s). Intermediate steps of rising were always performed before each reagent addition using Tris-HNO_3 buffers (Alves, Pimentel, Nouws, Correr, et al., 2015). Analyses were performed in triplicate and measurements were carried out at controlled temperature ($21 \pm 1^\circ\text{C}$).

2.4. Samples

A complex chocolate-based product containing peanut (20%) was acquired in a local supermarket. The sample also contained milk chocolate, sugar, milk powder and milk proteins, cocoa butter, palm fat, milk fat, starch, glucose, soya lecithin, coconut oil, salt, stabilizing agents, and flavor additives. As negative control, a commercial chocolate without peanut as ingredient (and without any labeled allegation of its potential presence) was used. It also contained the following ingredients: sugar, whole milk powder, skimmed milk powder, cocoa butter, cocoa paste, soya lecithin, palm fat, butter, and flavor additives.

2.5. Sample preparation

Samples were first frozen at -20°C (to avoid melting during milling) and then immediately ground (GM 200, Retsch, Germany) at 10,000 rpm during 20 s (3x). Different extracts were prepared, in triplicate, using a rigorously weighted amount ($\sim 2.00\text{ g}$) of powdered sample and $20.00 \pm 0.04\text{ ml}$ of extraction solution (Tris- HNO_3 buffer 0.1 M (A), $\text{pH} = 8$), with or without the addition of different amounts of NaCl (B) and/or skimmed milk powder (C). In summary: A) Tris- HNO_3 buffer ($\text{pH} = 8$), B) Addition of NaCl (0.1 or 1 M), C) Addition of skimmed milk powder (1 or 2 g). The extracts were obtained at controlled time, 15 or 60 min, and temperatures, $21 \pm 1^{\circ}\text{C}$ and $60 \pm 1^{\circ}\text{C}$ (E5CC, Omron Corporation, Kyoto, Japan), under constant magnetic stirring. Subsequently, the samples were subjected to a first centrifugation step (5000 rpm) during 5 min (Labofuge Ae, Heraeus Sepatech, Germany). An aliquot (1 ml) of the supernatant was further centrifuged at 10,000 rpm for 3 min (Heraeus Fresco 17 Centrifuge, Thermo Fisher Scientific, Germany) and the resulting supernatant was diluted (1:250) and used in the electrochemical immunoassay. Each extract was analysed in triplicate.

2.6. Statistical analysis

Statistical analysis was performed using IBM SPSS v. 20 (IBM Corp., Armonk, 241 NY, USA). Data are expressed as mean \pm standard deviation. The Shapiro-Wilk test was used to evaluate data normality. The one-way ANOVA was used to assess significant differences between samples, followed by Tukey's HSD or Dunnett T3 post hoc test (selected based on the equality of the variances) to make pairwise comparisons between means. The level of significance for all hypothesis tests (p) was 0.05. A hierarchical Cluster analysis was performed using the method of complete linkage (farthest neighbor) with Euclidean distances.

3. Results and discussion

Different sample preparation protocols have been used by several authors to perform biosensing detection of different food allergens. Extraction with a $\text{pH} 7\text{--}9$ buffer (by vortexing or in an ultrasonic bath) followed by centrifugation is a common procedure (Alves, Pimentel, Nouws, Marques, et al., 2015; Lu, Oshima, & Ushio, 2004; Maier, Morgan, Lindner, & Pittner, 2008; Yman, Eriksson, Johansson, & Hellenas, 2006). Other researchers preferred a commercial allergen extraction buffer (RIDASCREEN[®], R-Biopharm AG, Darmstadt, Germany) (Bremer, Smits, & Haasnoot, 2009; Raz, Liu, Norde, & Bremer, 2010). When food products are rich in phenolic compounds, which can bind to allergens and antibodies, skimmed milk powder can be added to eliminate their interference (Pollet et al., 2011). In some cases, the use of KCl or NaCl is also employed to improve extraction (Wang et al., 2011; Yman et al., 2006). In general, these protocols not only vary according to the sample but also according to the target analyte.

In what concerns to the commercial Enzyme-Linked Immunosorbent Assays (ELISA), several kits from different suppliers are available to detect major peanut allergens (Jayasena et al., 2015) consider their comparison difficult and unreliable due to the fact that they differ widely from each other in sample extraction procedures, range of quantification, limits of detection and quantification, reference standards used, and results expression. In fact, in a previous study (Zeleny and Schimmel (2010)) concluded that the extraction buffer influences the protein yield of extraction. In accordance, Poms, Capelletti, et al. (2004) showed that extraction buffers operating in the pH range 8–11 showed the best

yields and that elevated roasting temperatures in food processing reduce those values.

In order to evaluate the influence of several parameters on the Ara h 6 extraction and detection, different sets of time and temperature were tested, as well as different amounts of the additives (skimmed milk powder and NaCl), which are described in Table 1. The additives were selected based on their chemical properties: skimmed milk powder has the ability to block phenolic compounds of the matrix (Pomés et al., 2004) and the presence of NaCl modulates the ionic strength of the solution, which can influence protein solubility (Zayas, 1997).

First, a calibration curve was prepared using the purified Ara h 6 protein and the optimal dilution of the sample extract was studied using two of the extracts described in Table 1: A (21°C , 15 min) and A/1B/2C (60°C , 60 min). These were selected because they were obtained using the most 'extreme' conditions. Different dilutions of each extract were then made using the Tris- HNO_3 buffer ($\text{pH} 7$), namely 1:50, 1:100, 1:250, 1:500, 1:1000, 1:2500, and 1:5000. The 1:250 dilution was selected for all the other analyses, since the peak current intensity (i_p , μA) obtained for both extracts were slightly below to the intermediate point of the calibration curve. This is important since above a certain concentration (saturation level), in this case 100 ng/ml, the sensor will give similar responses that correspond to the maximum i_p value, and the comparison of the results would be impaired. A linear semi-logarithmic relationship between Ara h 6 concentration ([Ara h 6]) and the i_p was obtained between 1 and 100 ng/ml:

$$i_p (\mu\text{A}) = 12.47 \times [\text{Ara h 6}] - 2.7982, \quad r = 0.9995.$$

In order to study the influence of temperature and time on Ara h 6 extraction, the extractions were performed at $21 \pm 1^{\circ}\text{C}$ and $60 \pm 1^{\circ}\text{C}$, both during 15 and 60 min. For each set of time and temperature, different amounts of the two additives (NaCl and skimmed milk powder) were also included (Table 1), in a total of 36 different extraction conditions. Table 1 also shows the Ara h 6 concentration ($\mu\text{g/g}$ of sample) that was determined using each extraction procedure. As can be observed, the results were variable, with average values ranging from $39.32 \pm 6.50\text{ }\mu\text{g/g}$ (extract 1: A (21°C , 15 min), using only the Tris- HNO_3 buffer ($\text{pH} = 8$) as extraction solution) to $146.31 \pm 27.05\text{ }\mu\text{g/g}$ (extract 17: A/1B/1C (21°C , 60 min), using Tris- HNO_3 buffer in the presence of 1 M NaCl and 1 g of skimmed milk powder). These maximum values are in accordance to the expected range of Ara h 6 amounts described in literature, based on the peanut amount of the sample (Chen, Wang, El-Mezayen, Zhuang, & Dreskin, 2013; Koppelman et al., 2001; Misra, 2001).

For a better comparison and comprehension of the results, the quantified amounts were grouped and statistically analysed (Figs. 1 and 2).

Fig. 1–I shows the influence of time on the extracted amount of Ara h 6 when the extraction was performed at 21°C , using the different extraction mixtures. In general, a time increase from 15 to 60 min maintained or raised the extracted amount of Ara h 6. Although a general increase tendency was found (certainly due to the increase of the sample/solution contact time), only with the mixture A/2C, which was composed of 2 g of skimmed milk powder and Tris- HNO_3 buffer, the increase was statistically significant ($p < 0.05$). It is clear that the use of some additive proportions was beneficial for Ara h 6 extraction. Indeed, approximately 3-fold higher Ara h 6 amounts were obtained for both extraction times when mixtures A/0.1B/2C (containing 0.1 M NaCl and 2 g of skimmed milk powder) and A/1B/1C (containing 1 M NaCl and 1 g of skimmed milk powder) were used, compared to the use of the Tris- HNO_3 buffer alone. This is in accordance with a previous study of another peanut allergen (Ara h 1), in which it was noticed

Table 1

Variation of the Ara h 6 concentration ($\mu\text{g/g}$) in the commercial chocolate sample containing peanut according to the extraction conditions used.

Extract (n °)	Extraction conditions (code)*	Ara h 6 concentration($\mu\text{g/g}$ sample)	CV (%)	Extract (n °)	Extraction conditions (code)*	Ara h 6 concentration ($\mu\text{g/g}$ sample)	CV (%)
1	A (21 °C, 15 min)	39.32 \pm 6.50	16.5	19	A (60 °C, 15 min)	48.31 \pm 15.21	31.5
2	A/0.1B (21 °C, 15 min)	48.82 \pm 2.96	6.1	20	A/0.1B (60 °C, 15 min)	53.03 \pm 11.98	22.6
3	A/1B (21 °C, 15 min)	58.79 \pm 8.41	14.3	21	A/1B (60 °C, 15 min)	96.64 \pm 1.77	1.8
4	A/1C (21 °C, 15 min)	79.40 \pm 4.37	5.5	22	A/1C (60 °C, 15 min)	116.38 \pm 2.77	2.4
5	A/2C (21 °C, 15 min)	53.97 \pm 9.33	17.3	23	A/2C (60 °C, 15 min)	109.05 \pm 17.96	16.5
6	A/0.1B/1C (21 °C, 15 min)	54.15 \pm 11.02	20.4	24	A/0.1B/1C (60 °C, 15 min)	69.12 \pm 0.35	0.5
7	A/0.1B/2C (21 °C, 15 min)	121.94 \pm 19.42	15.9	25	A/0.1B/2C (60 °C, 15 min)	95.85 \pm 15.56	16.2
8	A/1B/1C (21 °C, 15 min)	105.74 \pm 19.20	18.2	26	A/1B/1C (60 °C, 15 min)	60.28 \pm 11.43	19.0
9	A/1B/2C (21 °C, 15 min)	82.87 \pm 13.59	16.4	27	A/1B/2C (60 °C, 15 min)	87.31 \pm 13.50	15.5
10	A (21 °C, 60 min)	48.20 \pm 4.69	9.7	28	A (60 °C, 60 min)	47.43 \pm 5.71	12.0
11	A/0.1B (21 °C, 60 min)	55.09 \pm 12.24	22.2	29	A/0.1B (60 °C, 60 min)	50.54 \pm 5.47	10.8
12	A/1B (21 °C, 60 min)	51.22 \pm 8.87	17.3	30	A/1B (60 °C, 60 min)	39.09 \pm 6.80	17.4
13	A/1C (21 °C, 60 min)	88.60 \pm 10.40	11.7	31	A/1C (60 °C, 60 min)	73.28 \pm 14.14	19.3
14	A/2C (21 °C, 60 min)	105.76 \pm 14.57	13.8	32	A/2C (60 °C, 60 min)	145.27 \pm 23.19	16.0
15	A/0.1B/1C (21 °C, 60 min)	52.52 \pm 11.33	21.6	33	A/0.1B/1C (60 °C, 60 min)	86.26 \pm 22.26	25.8
16	A/0.1B/2C (21 °C, 60 min)	145.86 \pm 24.03	16.5	34	A/0.1B/2C (60 °C, 60 min)	82.69 \pm 5.93	7.2
17	A/1B/1C (21 °C, 60 min)	146.31 \pm 27.05	18.5	35	A/1B/1C (60 °C, 60 min)	59.80 \pm 10.15	17.0
18	A/1B/2C (21 °C, 60 min)	99.12 \pm 2.20	2.2	36	A/1B/2C (60 °C, 60 min)	61.19 \pm 7.88	12.9

Mean \pm standard deviation, calculated analysing three extracts prepared in different days. Each extract was analysed in triplicate (total of replicates: $n = 9$).

* A, only Tris-HNO₃ buffer (pH = 8); A/0.1B, Tris-HNO₃ with 0.1 M NaCl; A/1B, Tris-HNO₃ with 1 M NaCl; A/1C Tris-HNO₃ with 1 g of skimmed milk powder; A/2C Tris-HNO₃ with 2 g of skimmed milk powder; A/0.1B/1C Tris-HNO₃ with 0.1 M NaCl and 1 g of skimmed milk powder; A/0.1B/2C, Tris-HNO₃ with 0.1 M NaCl and 2 g of skimmed milk powder; A/1B/1C, Tris-HNO₃ with 1 M NaCl and 1 g of skimmed milk powder; A/1B/2C, Tris-HNO₃ with 1 M NaCl and 2 g of skimmed milk powder; CV, coefficient of variation (%).

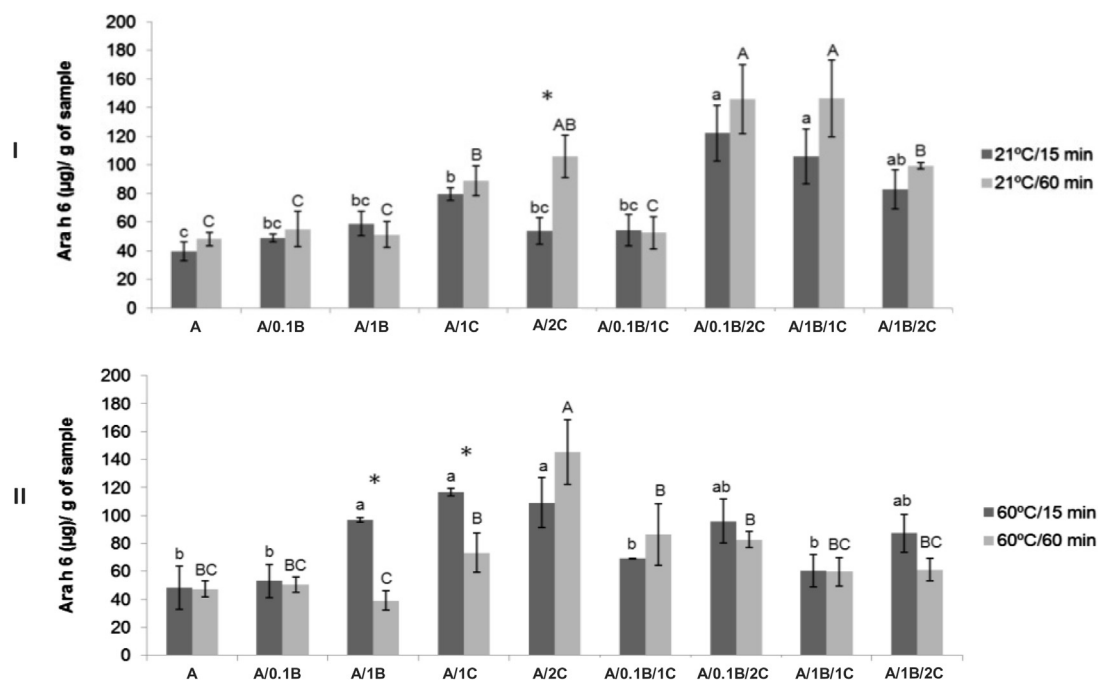


Fig. 1. Influence of extraction time (15 and 60 min) on the Ara h 6 amount (μg) extracted from a chocolate commercial sample containing peanut at 21°C (I) and at 60°C (II). A, only Tris-HNO₃ buffer (pH = 8); A/0.1B, Tris-HNO₃ with 0.1 M NaCl; A/1B, Tris-HNO₃ with 1 M NaCl; A/1C Tris-HNO₃ with 1 g of skimmed milk powder; A/2C Tris-HNO₃ with 2 g of skimmed milk powder; A/0.1B/1C Tris-HNO₃ with 0.1 M NaCl and 1 g of skimmed milk powder; A/0.1B/2C, Tris-HNO₃ with 0.1 M NaCl and 2 g of skimmed milk powder; A/1B/1C, Tris-HNO₃ with 1 M NaCl and 1 g of skimmed milk powder; A/1B/2C, Tris-HNO₃ with 1 M NaCl and 2 g of skimmed milk powder. * Indicates significant differences ($p < 0.05$) between 21°C/15 min and 21°C/60 min (I) or between 60°C/15 min and 60°C/60 min (II), for the same extraction mixture. Different small letters above left columns indicate significant differences $p < 0.05$ between different extraction mixtures used at 21°C/15 min (I) or 60°C/15 min (II). Different capital letters above right columns indicate significant differences $p < 0.05$ between different extraction mixtures used at 21°C/60 min (I) or 60°C/60 min (II).

that chocolate impaired its detection when foods were extracted without additives (Pomés et al., 2003). The authors concluded that this allergen should be extracted from chocolate products in the presence 5% of nonfat dry milk (Pomés et al., 2004).

By comparing the 15-min extractions at both temperatures (Fig. 2-I), in general, fast 'hot' extractions were better when using

only one additive: significantly higher results ($p < 0.05$) were obtained at 60°C for three extraction mixtures, namely A/1B (inclusion of 1 M NaCl), A/1C (addition of 1 g of skimmed milk powder) and A/2C (addition of 2 g of skimmed milk powder). In contrast, when using A/0.1B/2C (combination of 0.1 M NaCl and 2 g of skimmed milk powder) and A/1B/1C (combination of 1 M

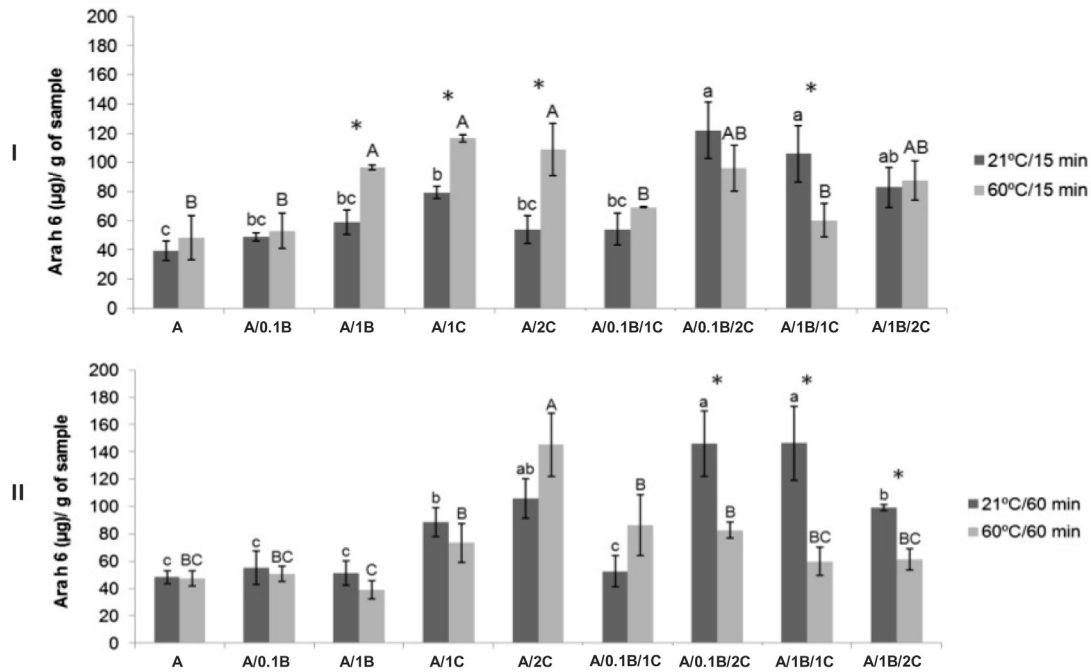


Fig. 2. Influence of extraction temperature (21 and 60°C) on the Ara h 6 amount (µg) extracted during 15 min (I) and 60 min (II) from the commercial chocolate sample containing peanut. **A**, only Tris-HNO₃ buffer (pH = 8); **A/0.1B**, Tris-HNO₃ with 0.1 M NaCl; **A/1B**, Tris-HNO₃ with 1 M NaCl; **A/1C**, Tris-HNO₃ with 1 g of skimmed milk powder; **A/2C**, Tris-HNO₃ with 2 g of skimmed milk powder; **A/0.1B/1C**, Tris-HNO₃ with 0.1 M NaCl and 1 g of skimmed milk powder; **A/0.1B/2C**, Tris-HNO₃ with 0.1 M NaCl and 2 g of skimmed milk powder; **A/1B/1C**, Tris-HNO₃ with 1 M NaCl and 1 g of skimmed milk powder; **A/1B/2C**, Tris-HNO₃ with 1 M NaCl and 2 g of skimmed milk powder. *Indicates significant differences ($p < 0.05$) between 21°C/15 min and 60°C/15 min (I) or 21°C/60 min and 60°C/60 min (II), for the same extraction mixture. Different small letters above left columns indicate significant differences ($p < 0.05$) between different extraction mixtures used at 21°C/15 min (I) or 21°C/60 min (II). Different capital letters above right columns indicate significant differences ($p < 0.05$) between different extraction mixtures used at 60°C/15 min (I) or 60°C/60 min (II).

NaCl and 1 g of skimmed milk powder), the obtained results were higher when the extractions were performed at room temperature. For all the remaining extractions, similar results were achieved using both temperatures, as detailed in Fig. 2-I.

For long extractions (60 min), comparing both temperatures (Fig. 2-II), significantly lower ($p < 0.05$) amounts of Ara h 6 were quantified when using the mixtures A/0.1B/2C (0.1 M NaCl and 2 g of skimmed milk powder), A/1B/1C (1 M NaCl and 1 g of skimmed milk powder), and A/1B/2C (1 M NaCl and 2 g of skimmed milk powder) at 60°C. Compared to extractions at 21°C, an increase of the average allergen amount (although not statistically significant) was obtained at the higher temperature, using the mixtures A/2C (2 g of skimmed milk powder) and A/0.1B/1C (0.1 M NaCl and 1 g of skimmed milk powder). For the remaining extractions, no significant differences were observed using different extraction temperatures Fig. 2-II).

A hierarchical clustering analysis, considering the extraction parameters as variables, showed that the operating conditions can be allocated into 7 main groups, according to the dendrogram presented in Fig. 3. In complete-linkage clustering (the methodology used), the link between two clusters contains all element pairs, and the distance between clusters equals the distance between those two elements (one in each cluster) that are farthest away from each other. The shortest of these links that remains at any step causes the fusion of the two clusters whose elements are involved (Yashwant & Sananse, 2015). The clusters obtained in Fig. 3 may be arranged, considering the decreasing order of Ara h 6 amount, as follows: Cluster 3 > Cluster 2 > Cluster 1 > Cluster 5 > Cluster 4 > Cluster 7 > Cluster 6.

Cluster 3 groups the set of conditions (no significant differences between them) that allowed the highest amount of Ara h 6 from the chocolate-based matrix. These are A/0.1B/2C (21°C, 60 min): 145.86 ± 24.03 µg/g; A/1B/1C (21°C, 60 min): 146.31 ± 27.05 µg/g;

g; and A/2C (60°C, 60 min): 145.27 ± 23.19 µg/g. Cluster 2 was composed of two sets of conditions – A/0.1B/2C (21°C, 15 min) and A/1C (60°C, 15 min) – that allowed the extraction of slightly lower amounts (121.94 ± 19.42 and 116.38 ± 2.77 µg/g, respectively). This was followed by Cluster 1 composed of six different sets of conditions, namely 8, 14, 18, 21, 23, and 25 (detailed in Table 1), in which Ara h 6 concentration varied from 95.85 ± 15.56 to 105.74 ± 19.20 µg/g and Cluster 5 (also constituted by six type of different extracts, Fig. 3) with minimum and maximum Ara h 6 amounts of 79.40 ± 4.37 and 88.60 ± 10.40 µg/g. Cluster 7 groups a great number of different extraction conditions (fifteen) that all gave Ara h 6 concentrations between 47.43 ± 5.71 and 61.19 ± 7.88 µg/g. Finally, the use of the following operating conditions – A (21°C, 15 min) and A/1B (60°C, 60 min) – showed to be the more inefficient to extract Ara h 6 from chocolate since in both cases less than 40 µg/g were detected.

Although the capacity of this sensor to detect and quantify Ara h 6 in real complex food samples was already tested and validated in a previous work (Alves, Pimentel, Nouws, Correr, et al., 2015), it was also important to find if the extraction mixtures selected as the best ones (Cluster 3, Fig. 3) could in some way interfere with the protein detection. Thus, blank extracts (without sample) were prepared and analysed showing similar results (no significant differences, $p > 0.05$) compared to the usual blank of the assay (using only Tris-HNO₃ buffer). Based on this, any possibility of interference of these extraction mixtures in this immunosensing detection can be excluded.

Besides, for the best extraction conditions (17), recoveries were also evaluated. For that, sample aliquots were spiked with 25, 50 or 75% of the expected Ara h 6 amount (standard). Then, spiked and non-spiked aliquots were extracted using those conditions. Extracts were analysed and recoveries calculated. For all the three levels, recoveries were similar or higher than 94.7%.

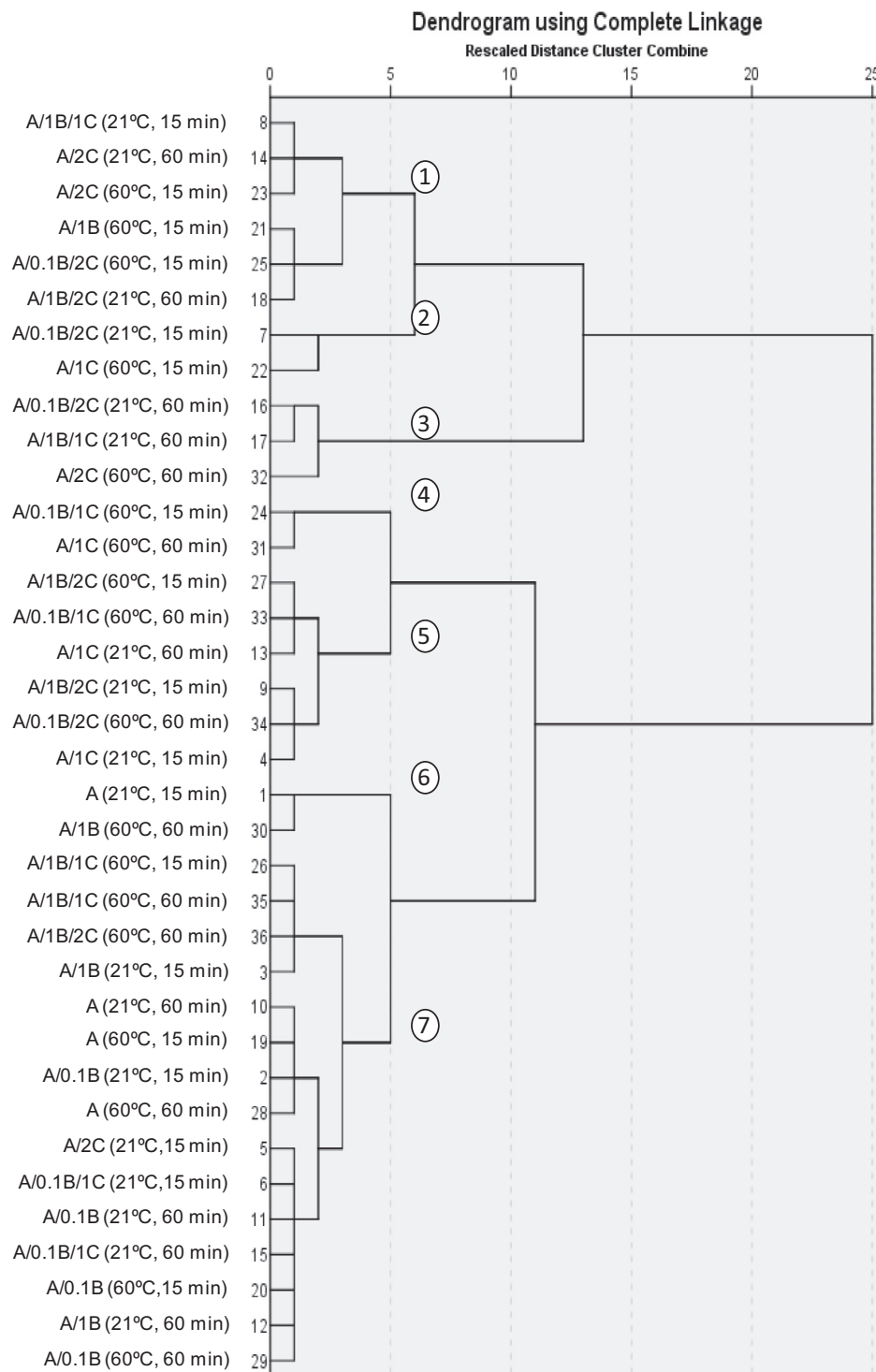


Fig. 3. Dendrogram resulting from a cluster analysis of the different extraction conditions tested. The set of conditions that correspond to each extract number are detailed in [Table 1](#).

4. Conclusions

In this work, different extraction conditions were tested in order to increase the extractability and detection of Ara h 6. The results of this work show that the use of additives can significantly increase Ara h 6 extraction, and that long extractions (60 min) are generally more effective than short ones (15 min), except in some of the cases (at 60°C), in which the use of this temperature

resulted in a decrease of the amount of extracted protein. Based on the cluster analysis, three sets of conditions were selected as the best ones since they allowed the extraction of the highest amounts of Ara h 6. This work highlights the importance of sample preparation and the need to adjust extraction procedures, taking into consideration not only the target analyte but also the components of the food matrix. This is a relevant issue regarding those commercial immunoassay kits that contain only one extraction

buffer/solution to perform food allergen extraction from different food samples.

Conflict of interest

The authors state that there are no conflicts of interest.

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